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sulphate ion is an anion derived from the sulphuric acid in which the valency of the hydrogen atom is one as hydrogen is more electropositive than sulphate 1 will be its oxidation number

may 19 2024 this table of element valences includes the maximum valence and most common valence values in

chemistry use this for reference with a periodic table

what is the valency of sulphate so 42 answer valency refers to the number of electrons that are gained or lost by an atom in order to complete its outermost shell

sep 13 2025 in case of the sulphate ion the valency is 2 because the charge on the ion is 2 this implies that sulphate has the ability to bind with two positive ions cations to form neutral

the document provides a valency chart for writing molecular formulas of chemical compounds categorizing ions and radicals by their oxidation states it lists positive ions monovalent

may 25 2014 it has a valence of 1 usually its oxidation state is 1 but it can lose the electron and have a valence of 1 the most stable oxidation state is one that fills or half fills an atom's

the valency of sulphate ion is determined by the number of electrons it can gain lose or share in a chemical reaction sulphate ion has six valence electrons and it requires two more electrons

number of the hydrogen atom two atoms of hydrogen 2h combination with element radical one radical of sulphate SO_4 compound formed sulphuric acid H_2SO_4 valency

sulphate SO_4^{2-} sulphate is a polyatomic ion consisting of 1 sulfur atom covalently bonded to 4 oxygen atoms it carries an overall charge of 2 the valency of sulphate is the charge it

valency of sulphate ion is 2 as two negative ions are found after the formation of sulphate ions which get formed due to sulphur and oxygen with sulphate valency of 2 it reacts with positive

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